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## WHAT IS CLAIMED IS:

1. An optical glass for polarizing optical system having a photoelastic constant C in the range of -0.2 to +0.5 [ $10^{-8}$  cm<sup>2</sup>/N] with respect to a wavelength of 633 nm, the optical glass having the following composition (1):

composition (1): when represented in terms of wt.% of oxides:

 $SiO_2$ : 17.0 - 27.0 % (35.5 - 57.0 mol%)  $Li_2O$  + Na<sub>2</sub>O + K<sub>2</sub>O: 0.5 - 5.0 % (0.7 - 20.0 mol%) PbO: 72.0 - 75.0 % (39.1 - 45.0 mol%)  $As_2O_3$  +  $Sb_2O_3$ : 0.1 - 3.0 % (0.1 - 2.0 mol%).

2. An optical glass for polarizing optical system having a photoelastic constant C in the range of -0.2 to  $+0.5 \ [10^{-8} \ cm^2/N]$  with respect to a wavelength of 633 nm, the optical glass having the following composition (2):

composition (2): when represented in terms of mol%:

 $SiO_2$ : 40.0 - 54.0 mol%

R<sub>2</sub>O (R: alkali metal): 0.5 - 9.0 mol%

PbO: 43.0 - 45.5 mol%

 $As_2O_3 + Sb_2O_3$ : 0.1 - 1.5 mol%; and

the composition (2) further containing fluorine in the following range when represented in terms of mol%:

fluorine/oxygen (F/O) ratio: 0.1 - 18.0.

3. An optical glass for polarizing optical system

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having a photoelastic constant C in the range of -0.2 to  $+0.5 \ [10^{-8} \ cm^2/N]$  with respect to a wavelength of 633 nm, the optical glass having the following composition (3):

composition (3) when represented in terms of mol%:

 $SiO_2$ : 40.0 - 54.0 mol%

R,0 (R: alkali metal) / 0.5 - 9.0 mol%

RF: 0 - 16.0 mol%

 $R_2SiF_6: 0 - 3.3 \text{ mol}$ %

PbO + PbF<sub>2</sub>: 43.0 - 45.5 mol%

PbF<sub>2</sub>:  $0 - 10.0 \text{ mol}_{8}^{1}$ 

 $As_2O_3 + Sb_2O_3$ : 0.1 /- 1.5 mol%; and

the composition (3) further containing fluorine in the following range in terms of mol%:

fluorine/oxygen (F/O) ratio: 0.1 - 18.0.

4. A process for producing an optical glass for polarizing optical system, the process comprising:

changing the ratio of PbO in a lead-containing optical glass to control the photoelastic constant C thereof to provide an optical glass for polarizing optical system having a photoelastic constant C in the range of -0.2 to +0.5 [ $10^{-8}$  cm<sup>2</sup>/N] with respect to a wavelength of 633 nm,

the optical glass having the following composition (1):

composition (1): when represented in terms of wt.% of oxides:

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SiO_2: 17.0 - 27.0 % (35.5 - 57.0 mol%)

Li_2O + Na_2O + K_2O: 0.5 - 5.0 % (0.7 - 20.0 mol%)

PbO: 72.0 - 75.0 % (39.1 - 45.0 mol%)

As_2O_3 + Sb_2O_3: 0.1 - 3.0 % (0.1 - 2.0 mol%).
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13 14 15 5. A process for producing an optical glass for polarizing optical system, the process comprising:

changing the fluorine/oxygen (F/O) ratio of a fluorine-containing optical glass so as to regulate the refractive index thereof while retaining the photoelastic constant C of the optical glass in the range of substantially zero to provide an optical glass for polarizing optical system having a photoelastic constant C in the range of -0.2 to +0.5 [10<sup>-8</sup> cm<sup>2</sup>/N] with respect to a wavelength of 633 nm,

the optical glass having the following composition (2):

composition (2): when represented in terms of mol%:

SiO<sub>2</sub>: 40.0 - 54.0 mol%

 $R_2O$  (R: alkali metal): 0.5  $\frac{1}{1}$  9.0 mol%

PbO: 43.0 - 45.5 mol%

 $As_2O_3 + Sb_2O_3$ : 0.1 - 1.5 mo½; and

the composition (2) further containing fluorine in the following range when represented in terms of mol%:

fluorine/oxygen (F/O) ratio: 0.1 - 18.0.

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6. A process for producing an optical glass for

polarizing optical system, the process comprising: changing the fluorine/oxygen (F/O/) ratio of a fluorine-containing optical glass so/as to regulate the refractive index thereof while retaining the photoelastic constant C of the optical glass in the range of substantially 5 zero to provide an optical glass for polarizing optical system having a photoelastic constant C in the range of -0.2 to  $+0.5 [10^{-8} \text{ cm}^2/\text{N}]$  with respect to a wavelength of 633 nm, the optical glass having the following composition Į. [] []10 (3): 'n composition (3) when represented in terms of mol%:  $SiO_2$ : 40.0 - 54.0 mol% tħ  $R_{2}0$  (R: alkali metal): 0.5 - 9.0 mol% Į. 13 13 15 RF: 0 - 16.0 mol%  $R_2SiF_6: 0 - 3.3 mol$ % PbO + PbF<sub>2</sub>: 43.0 - 45.5 mol%  $PbF_{2}: 0 - 10.0 \text{ mol}$ % As,0, + Sb,0,: 0.1 - 1.5 mol%; and the composition (3) further containing fluorine in the 20 following range in terms  $\phi f$  mol%: fluorine/oxygen (F/O) ratio: 0.1 - 18.0.